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6.2 Covalent Bonding and Molecular Compounds **Ionic and Covalent Bonds Made Easy**

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 Chemical Bonds - Covalent Bonds And  
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 ANSWER** Answer the following questions  
 in the space provided. 1. a A chemical

bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.6 Chemical BondingChapter 6 Review: Chemical Bonding. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. hanstep. Key Concepts: Terms in this set (40) A chemical bond between atoms results from the attraction between the valence electrons and \_\_\_\_ of different atoms. nuclei. A covalent bond consists of.Chapter 6 Review: Chemical Bonding Flashcards | QuizletChapter 6 Review: Chemical Bonding. 40 terms. alifay16. study review sheet (chapter 8 test) 25 terms. jessg202. Bio 111 CLOs Basic Chemistry.

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in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear molecule?6  
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#### SECTION 6-2 SHORT ANSWER Answer

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an attraction between atoms that allows the formation of chemical substances that contain two or more atoms. The bond is caused by the electrostatic force of attraction between opposite charges, either between electrons and nuclei, or as the result of a dipole attraction.

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