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 HBr 4 NH₄OH - KAl (SO₄)₂ · 12H₂O - Al (OH)₃ (NH₄)₂SO₄ - KOH 12 H₂O

When you balance the chemical equation, it's always a good idea, to check the final equation, to make sure it works. Check: Reduce the numbers of each type of atom. The total number of atoms in the balanced equation will be the same on both sides of the equation.

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- How many atoms are in a hydrogen molecule (H₂)? 2
- How many atoms are in an oxygen molecule (O₂)? 2
- How many hydrogen and oxygen atoms are in a water molecule (H₂O)? H: 2 O: 1
- In general, what does a subscript (such as the "2" in H₂) mean?

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Student Exploration: Balancing Chemical Equations Vocabulary: coefficient,

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Introduction: Chemical equations show how compounds and elements react with one another. An element is a substance consisting of one kind of atom, such as aluminum (Al) or

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Emma_-

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The chemical equation has the products on the right side, while the reactants are written on the left side. Both of them are separated by an arrow. For instance, $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ denotes that there are four atoms of hydrogen and 2 atoms of oxygen on both sides of the equation. The amount of reactants must be equal to the amount of products.

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