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CHAPTER 8 Chemical Equations and Reactions

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hand side connected by a one-headed or two-headed arrows. Chemicals Reactions - Definition, Equations, Types & Examples

- To balance a chemical equation, it is permissible to adjust the
 - coefficients.
 - subscripts.
 - formulas of the products.
 - number of products.
- In a chemical equation, the symbol (aq) indicates that the substance is
 - water.
 - dissolved in water.
 - an acid.
 - insoluble.
- The production of a slightly soluble solid compound in a double-displacement reaction results in the ...

Chapter 8: Chemical Equations and Reactions

A chemical equation also indicates the number of atoms of each element involved in a reaction.

8. Balanced Chemical Equations

There are three ways of writing a chemical equation:

- Word equation

The above chemical reaction between zinc and mineral acid can be represented as: Zinc + Hydrogen chloride → Zinc chloride + Hydrogen

In a word ...

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Chemical Reaction: During chemical reactions, the chemical composition of substances changes or new substances are formed.

- Chemical Equation: Chemical reactions can be written in chemical equation form which should always be balanced.
- Types of Chemical Reactions: Combination reaction: A single product is formed from two or more reactants.

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Chemical Equations and Reactions Section Quiz: Describing Chemical Reactions

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

___ b ___

- Which of the following is true regarding the law of conservation of mass?
 - It dictates that the number of molecules on each side of a chemical equation must be the same.

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Chemical Reaction - process in which one or more substances are changed into one or more different substances.

Reactants - original substance. Products - resulting substance . Described by Chemical Equations.

Chemical equation - represents, with symbols and formulas, the identities and molar amounts of the reactants and products in a ...

Chapter 8 - Chemical Equations and Reactions

Chapter 8: Chemical Equations and Reactions

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- 3.

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1. To balance a chemical equation, it is permissible to adjust the
a. coefficients. b. subscripts. c. formulas of the products. d.
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indicates that the substance is a. water. b. dissolved in water. c.
an acid. d. insoluble., 3. The production of a slightly soluble solid
compound in a double-displacement reaction results in the ...

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