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 properties using the
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 Periodicity:regular variations (or patterns) of properties with increasing atomic weight; both chemical and physical properties vary in a periodic (repeating pattern).
 Group:vertical column of elements ("family")
 Period:horizontal row of elements.
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highest electronegativity. *Periodic Table Trends Notes Lake K12 FI*
The metallic character of the elements decreases as you move across a Period on the Periodic table, from left to right, and it increases as you move down a Group This trend occurs due to atoms more readily accepting electrons to fill their valence shells rather than losing them to have the below, already full, electron shell as their outer shell

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The Periodic Table. The Periodic table is a method of classifying elements. They are arranged in order of increasing atomic number. Made up of rows called periods and columns called groups.

Metals are to the left and non-metals are to the right. Elements in the same group have similar properties.

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Periodic Classification of Elements, Class 10 Chapter 5 Science Notes includes detailed explanations about the modern periodic table, periodic trends and about its discovery. [Periodic Classification of Elements Class 10 Notes](#)

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