

# History Of Atomic Theory Webquest Answers

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History of Atomic Theory Webquest Part 1: Atom Basics Go to: and read the “And you thought you were strange” section to fill the table below. Subatomic Particles Location in the Atom Charge 1. protons center positive 2. electrons outside negative 3. neutrons center none Part 2: Early Ideas about Atoms Go to and read the section on “Early Greek Ideas” in order to answer the questions below.

### History of Atomic Theory Webquest

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DOC Atomic Model/Theory Webquest. History of Atomic Model Web-Quest Poster Project- Due October 1st, 2015. Learning Target: I can describe the importance of models for the study of atomic structure. I can describe the crucial contributions of scientists and the critical experiments that led to the development of the modern atomic model.

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Webquest: Atomic Theories and Models Answer these questions on your own, USING COMPLETE SENTENCES where appropriate (most of the questions, except tables and drawings).

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*WebQuest: The Atomic Theory: A History*

The following Webquest will help you review content covered throughout lecture and discussion. It will help you review topics such as: Atom Basics.

Early Ideas About Atoms. John Dalton's Atomic Theory. JJ Thomson & the Electron. Chadwick, Rutherford, and the Atomic Nucleus. History of the Atom

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Webquest: Atomic Theories and Models Answer these questions on your own, ... and what was the result on the progress of atomic theory ... Search the internet (and your worksheet) to fill in your History of the Atom Timeline. Use the following clues to help you.

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Dalton's atomic theory is based on the following postulates: 1.All matter consists of tiny particles 2.Atoms are indestructible and unchangeable

3.Elements are characterized by the mass of there atoms

#### **Answers To History Of Atomic Theory Webquest**

1. In 442 BC, who developed the Greek atomic theory? 2. What did this theory say? 3. For how long was this theory accepted? 4. What did Aristotle's new theory say? 5. Why was this theory so much more accepted by the public when his theory was way off? 6. Did Lavoisier, the father of modern chemistry, develop an atomic theory? 7. If so what did ...

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name the five postulates of the atomic theory. 1. all matter is composed of extremely small particles called atoms 2.all atoms of a given element are identical (same size,mass, and chemical properties).