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 proteins constructed (Figure 2.12). Fibrous, or structural, proteins
 are the basic structural materials of the body.Chapter 2: Basic
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 matter, made up of three subatomic particles. Protons have a
 positive charge and neutrons carry no charge. Strong forces bind
 protons and neutrons together in the nucleus.chapter_2.1 -
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 2 Objectives (IWBAT) The student evaluates the energy needs of
 the human body and processes through those needs 1.
 Differentiate matter from energy. 2. List four major energy forms,
 and provide one example of how each energy form is used in the
 body. 3. Define chemical element, and list the four elements that
 form the bulk of body matter.Chapter 2 Basic Chemistry -
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 The "stuff" of the universe. Anything that occupies space and has
 mass. States of matter Solid Liquid Gas Basic Chemistry Chapter
 2: Chemistry Comes Alive Energy Less tangible no mass, does not
 take up space, & is only measured by its effects on
 matter.Chemistry Comes Alivechapter 2 basic chemistry packet
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 8. Polarity in a water molecule is caused by an uneven
 distribution of electrons between the oxygen and hydrogen
 atoms. 9. The concentration of H ions determines whether a
 solution is acidic or basic. 10. Capillary action is the effect of
 water rising in a narrow tube against the force of ...Ch. 2 Answer
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 is anything that takes up space and has mass energy is the
 capacity to do work or put matter into motionCHAPTER 2 Basic
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 2 - Basic Chemistry 1. The Chemistry of LifeChapter 2 2.
 Chemistry All living things depend on chemistry The structure
 and function of all living things are governed by the laws of

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