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empirical formula = CH₂ [1] b mass of C in CO₂ = $12.01 \frac{44.01}{44.01} = 0.830$ g [1] mass of H in H₂O = $2.02 \frac{18.02}{18.02} = 0.139$ g [1] mass of O = $1.52 - (0.830 + 0.139) = 0.551$ g [1]

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