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As the temperature is increased, the average kinetic energy increases and more particles have enough kinetic energy to overcome the forces keeping them in the liquid state. Name Date Class STATES OF MATTER 13 Chapter 13 States of Matter pages 341 to 362. Properties of fluids. Gases and liquids are both fluids. Both these states of matter have greater freedom of motion. Objects exert pressure. Pressure... Chapter 13 States of Matter notes - callaghan13 STUDY GUIDE FOR CONTENT MASTERY CHAPTER States of Matter Section 13.1 Gases In your textbook, read about the kinetic-molecular theory. Complete each statement. 1. The kinetic molecular theory describes the behavior of gases in terms of particles in 2. The kinetic-molecular theory makes the following assumptions. a. CHAPTER 13 STATES OF MATTER.pdf Chapter 13 - States of Matter Chapter 14 - Behavior of Gases Chapter 15 - Water and Aqueous Systems Chapter 16 - Solutions Chapter 17 - Thermochemistry Chapter 18 - Reaction Rates and Equilibrium Chapter 19 - Acids, Bases and Salts Chapter 20 - Oxidation-Reduction Reactions Chapter 13 - States of Matter Chapter 13 "States of Matter". glass

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