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Chapter 8: Covalent Bonding. Matter takes many forms in nature: In this chapter, we are going to learn to distinguish the type of compound that we have already studied, the “ionic compound” (which contains oppositely-charged particles: metal cations and non-metal anions), from a different type of compound – a “molecular compound”.

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Nonpolar Covalent Bond Character

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called a coordinate covalent bond because

a. both bonding electrons come from the oxygen atom. b. it forms an especially strong bond. c. the electrons are equally shared. d. the oxygen no longer has eight valence electrons.

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