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Chemistry 213 Assignment No. 2 Name: Summer 2010

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QUESTIONS Select the one best answer for each question. 1. In

the reaction  $\text{CaCO}_3 + 2\text{HNO}_3 \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{CO}_2 + \text{H}_2\text{O}$ , how

many milliliters of 0.1015 M  $\text{HNO}_3$  are required to react with

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where X is the halogen. The barium content of the salt can be

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$8.6 \times 10^{-4}$  M.  $\text{SrF}_2$   $\text{Sr}^{2+} + 2\text{F}^-$   $K_{sp} = [\text{Sr}^{2+}][\text{F}^-]^2 = [\text{Sr}^{2+}][8.6 \times 10^{-4}]^2 = 2 \times 8.6 \times 10^{-4} = 17.2 \times 10^{-4}$   $K_{sp} = 8.6 \times 10^{-4}$

$4 \times 10^{-4} = 1.5 \times 10^{-4}$  2. Calcium fluoride,  $\text{CaF}_2$ , is a sparingly

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