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b. properties. d. symbols. ___7.

Magnesium (Mg) is located to the right of sodium (Na) because Mg has a. fewer protons. c. no protons. b. no neutrons. d. more protons. ___8. As you move from left to right across the periodic table, elements a. become less metallic. c. have a lower atomic weight. b. have a lower atomic number. d. become more metallic. ___9.

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B) Atoms of different elements have different properties. C) The volumes of

gases that combine are in small whole number ratios. D) Atoms combine in fixed ratios of whole numbers when they form compounds. E) In chemical reactions, atoms are not created or destroyed. Ans: C Introduction to Chemistry 3rd Edition Bauer Test Bank

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B)chlorine is more metallic than sodium
C)chlorine has a greater ionization energy than sodium does
D)chlorine has a greater electron affinity than sodium does
E)chlorine is bigger than sodium 46) Of the elements below, _____ is the most metallic.

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Electrons removed from atoms always

increase the stability of the atom. The energy of electrons decreases with increasing distance from the nucleus. Electrons are considered valence electrons if...

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