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Transparency Masters Chemistry: Matter and Change • Chapter

11 31 number of moles of known substance C H C H C H O O O CO

CO CO H O H O H O number of moles of unknown substance O CO

H O C H CO H O C H O H O C H O CO use mole ratio 5 mol O 2 1

mol C 3 H 8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1 mol ...MATH

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chemical equation at the top of the transparency.Math Skills

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WORKSHEET Determining Numbers of Ions Use with Chapter 7, Section 7.3 1. Write a simple word equation that illustrates what must be true of total positive charge and total negative charge in an ionic compound. 2. How many potassium ions (group 1) would be related to balance the charge of each

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Name Date Class Name Date Class MATH SKILLS TRANSPARENCY

WORKSHEET 1 MATH SKILLS TRANSPARENCY WORKSHEET 2

Interpreting and Drawing Graphs Use with Chapter 2, Visualizing

the Use with Chapter 3, Section 2.4 Conservation of Mass Section

3.2 1.MATH SKILLS TRANSPARENCY MASTER Visualizing the ...

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charge and total negative charge in an ionic compound. 2. How many potassium ions (group 1) would be related to balance the charge of each of

**Math Skills Transparency Worksheet Answers Chapter 3 ...** MATH SKILLS TRANSPARENCY WORKSHEET Use with Chapter 25, Section 25.3 42 82 1. In equation 1, what do the numbers 238 and 92, written to the left of the symbol U, represent? 238 is the mass number of the isotope, and 92 is the atomic number of the isotope. 2. Explain how both mass number and atomic number have been conserved in equation 1.

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Name Date Class Use with Chapter 6, Section 6.2

MATHSKILLSTRANSARENCY WORKSHEET Using the Periodic Table 1. Identify the number of valence electrons in each of the following elements.

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94 Chemistry: Matter and Change • Chapter 4 Math Skills Transparency Masters Calculating Atomic Mass MATH SKILLS

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