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# Chemistry Chapter 10 Chemical Quantities

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 Quantities. STUDY. PLAY. mole.  $6.02 \times 10^{23}$   
 representative particles of a  
 substance, SI unit for measuring the  
 amount of a substance. Avogadro's  
 number.  $6.02 \times 10^{23}$  representative  
 particles of a substance, named in honor  
 of Italian scientist Amedeo Avogadro di  
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 measure matter? If it is something large  
 like the number of beans in a bag, you just  
 count them. If, however, it is something  
 exceedingly small like an atom or ion, how  
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10. Look at Figure 10.16 and Table 10.3.  
 Name three compounds that have an  
 empirical formula of CH. 11. Fill in the  
 labels on the diagram below.

MICROSCOPIC INTERPRETATION  
 MACROSCOPIC INTERPRETATION SO<sub>3</sub>  
 molecule 1 mol SO<sub>3</sub> SO<sub>3</sub> composed of  
 composed of S atom and sulfur atoms (  $10^{23}$ )  
 oxygen atoms 3 atoms and  
 CHAPTER 10,Chemical  
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