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7 REVIEW

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SECTION 3

SHORT

ANSWER

Answer the

following

questions in

the space

provided. 1.

Label each of

the following

statements as

True or False:

True a. If the

formula mass

of one

molecule is x

amu, the

molar mass is

x g/mol. False

b. Samples of

equal

numbers of

moles of two

different

chemicals7

Chemical

Formulas and

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Compoundsch

emical

formula

reveals the

number of

atoms of each

element con-

tained in a

single

<p>molecule of the compound, as shown below for the hydrocarbon octane. (Hydrocarbons are molecular compounds composed solely of carbon and hydrogen.)</p> <p>CHEMICAL FORMULAS AND CHEMICAL COMPOUNDS</p> <p>203 SECTION 7-1 OBJECTIVES</p> <p>Explain the significance of a chemical ...CHAPTER 7 Chemical Formulas and Chemical CompoundsChapter 7: Chemical</p>	<p>Formulas and Chemical Compounds</p> <p>Section 7-1: Chemical Names and Formulas</p> <p>7-1-1 Explain the significance of a chemical formula. The subscript indicates that there are 8 carbon atoms in the molecule of octane. The subscript indicates that there are 18 hydrogen atoms in the molecule of octane. Subscript 2 refers to 2 aluminum ...Chapter 7: Chemical Formulas and</p>	<p>Chemical CompoundsStart studying Chapter 7 Chemical Formulas. Learn vocabulary, terms, and more with flashcards, games, and other study tools.Chapter 7 Chemical Formulas Flashcards QuizletStart studying Chapter 7: Chemical Formulas. Learn vocabulary, terms, and more with flashcards, games, and other study tools.Chapter 7: Chemical Formulas</p>
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sum of the positive charges equals the sum of the negative charges 2.	2H 6 = ethane (2 carbon atoms, 6 hydrogen atoms) B. Ionic Compounds	copper(II) carbonate Na ₂ SO ₃ b. sodium sulfite (NH ₄) ₃ PO ₄ c. ammonium phosphate
a.WORKSHEET 1: Determination of oxidation number or valence ...Chapter 7 - Chemical Formulas and Chemical Compounds 7-1 Chemical Names and Formulas I. Significance of a Chemical Formula A. Molecular formulas 1. Number of atoms of each element in one molecule of a compound C	1.Chapter 7 - Chemical Formulas and Chemical CompoundsCHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO ₃ a.	SnS 2 d. tin(IV) sulfide HNO ₂ e. nitrous acid 2.7 Chemical Formulas and Chemical CompoundsCHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the

<p>specified element in each of the following examples: a. S in H_2SO_3 b. S in $MgSO_4$ c. S in K_2S d. Cu in Cu_2S e. Cr in Na_2CrO_4 f. N in HNO_3 g. C in (HCO_3) h. N in $(NH_4)_2$.</p> <p>a. Chemical Formulas and Chemical Compounds - MAFIADOC.COM CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c</p>	<p>In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on ...7</p> <p>Chemical Formulas and Chemical Compounds Test and improve your knowledge of Holt McDougal Modern Chemistry Chapter 7: Chemical Formulas and Chemical</p>	<p>Compounds with fun multiple choice exams you can take online with Study.com Chemistry Chapter 7 Worksheets —Chemical Formulas and Nomenclature page 1 WORKSHEET 1: Determination of oxidation number or valence number Rules to apply: 1. a. The net charges on all molecules is zero; therefore, the sum of the positive charges equals the sum of the</p>
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number or valence ...

7. What is the empirical formula for lactic acid, which has a molecular formula of (331--1603? a. b. CHO 232 c. CHO 363 d. CHO 8. NaCl is a. a molecular formula only. b. an empirical formula only. c. both an empirical formula and a chemical formula. d. both a molecular formula and an empirical formula. 9.

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atoms of each	chools.com	ANSWER
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shown below	interactive	for the
for the	flashcards.	following
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octane.	500 different	CuCO ₃ a.
(Hydrocarbons	sets of	copper(II)
are molecular	chemistry	carbonate Na
compounds	chapter 7	2SO ₃ b.
com-posed	chemical	sodium sulfite
solely of	formulas	(NH ₄) ₃ PO ₄ c.
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SHORT	moles of two	Chemical
ANSWER	different	formulas.
Answer the	chemicals	Monatomic
following	<i>7 Chemical</i>	ions. C and Si.
questions in	<i>Formulas and</i>	Binary ionic
the space	<i>Chemical</i>	compound.
provided. 1.	<i>Compounds</i>	Show relative
Label each of	Choose from	number of
the following	500 different	atoms in a
statements as	sets of	chemical
True or False:	formula	compound.
True a. If the	chapter 7	ions formed
formula mass		from 1 atom.

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