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 Guidewhere $\hbar = h/2\pi$ and
 m is the mass of the
 electron. The integer n is
 known as a "quantum
 number" and illustrates
 the fundamental
 discreteness of quantum
 processes. Furthermore
 there must be a balance
 of forces in the orbit, the
 electrical attraction of the
 electron to the nucleus is
 equal to the centrifugal
 force $e^2/r^2 = mv^2/r$.
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(CM332C)Photons are
 particle-like quanta of
 electromagnetic radiation.
 They travel at the speed
 of light c with momentum
 p and energy E given by
 $p = h/\lambda$ and $E = hc/\lambda$, (1:1)
 where λ is the wavelength
 of the electromagnetic
 radiation. In comparison
 with macroscopic
 standards, the
 momentum and energy of
 a photon are
 tiny.INTRODUCTION TO
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 can be thought of roughly
 as the study of physics on
 very small length scales,

although there are also
 certain macroscopic
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Quantum mechanics can be thought of roughly as the study of physics on very small length scales, although there are also certain macroscopic systems it directly applies to. The descriptor "\quantum" arises because in contrast with classical mechanics, certain quantities take on only discrete values.

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