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# Chapter 6 Chemical Bonds Answers

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6. The two main types of chemical bonding are ionic and covalent bonding. Ionic bonds are formed when a transfer of electrons takes place.  
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fullexams.com • A covalent bond is a chemical bond in which two atoms share a pair of valence electrons. • When two atoms share one pair of electrons, the bond is called a single bond. When two atoms share two pairs of electrons, the bond is called a double bond. • A molecule is a neutral group of atoms that are joined together by one or more covalent bonds.  
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Name \_\_\_\_\_  
\_\_\_\_\_ each statement or best answers each question.  
1. If two covalently bonded atoms move closer than a distance of the bond length, the potential energy of the atoms a. becomes negative. b. decreases. c. increases. d. remains constant.  
2. The electrons involved in the formation of a covalent bond are a. transferred from one atom to another.  
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