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1? ... In your answer, consider the external and internal pressure data given in Model 1.

*Sam Schoderbek Chemistry: Gas Variables Pogil*

Explain your answer. 10, A scuba tank contains a mixture of oxygen and helium gases, Before the dive, the partial pressure of oxygen gas is 0.65 atm and the partial pressure of helium is 0.38 atn. a. What is the total pressure in the scuba tank before the dive? b. After the dive, the partial pressure due to the oxygen gas is reduced by 80%.

*Gavin Gutowsky Chemistry Blog: Gas Variables Pogil*

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