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SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)

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Match each item with the correct

statement below. a. calorimeter d.

enthalpy b. calorie e. specific heat c.

joule f. heat capacity ____ 1. quantity of

heat needed to raise the temperature of

1 g of water by 1°C ____ 2. SI unit of

energy ____ 3. Ch 17 Thermochemistry

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changes and Hess's Law. Chapter 17 - Thermochemistry - Mrs. Gingras' Chemistry Page Chapter 17 Thermochemistry 183 SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. Energy Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505–510) CHAPTER 17, Thermochemistry (continued) 2. What is calorimetry? Calorimetry is the accurate and precise measurement of heat change for chemical and physical processes. 3. Use Figure 17.5 on page 511. Circle the letter next to each

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Chapter 17 review Important Vocabulary

- Thermochemistry- Studies of energy changes that occur during chemical reactions and change state
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- Heat- Represented by q , is energy that transfers from one object to another

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Matching Match each item with the correct statement below.

a. calorimeter
d. enthalpy
b. calorie
e. specific heat
c. joule
f. heat capacity

___ 1. quantity of heat needed to raise the temperature of 1 g of water by 1°C
___ 2. SI unit of energy
___ 3.

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