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 following questions in the space provided. 1. b Pressure surf f a o  
 c r e ce area. For a constant force, when the surface area is  
 tripled the pressure is (a) doubled. (b) a third as much. (c) tripled.  
 (d) unchanged.mc06se cFMsR i-vi11.1 Gases and Their Properties  
 463 For an ideal gas (in which the particles occupy no volume  
 and experience no attractions or repulsions), gas pressure and  
 volume are inversely proportional.Chapter 11 Gases - An  
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 11 REVIEW Gases SECTION 4 SHORT ANSWER Answer the  
 following questions in the space provided. 1. \_\_\_\_ List the  
 following gases in order of increasing rate of effusion. (Assume all  
 gases are at the same temperature and pressure.) (a) He (b) Xe  
 (c) HCl (d) Cl<sub>2</sub> 2.CHAPTER 11 REVIEW Gases - Manasquan Public  
 SchoolsSection Goals and Introductions Section 11.1 Gases and  
 Their Properties Goals To describe the particle nature of both real  
 and ideal gases. To describe the properties of gases that can be  
 used to explain their characteristics: volume, number of particles,  
 temperature, and pressure.Chapter 11 - GasesChapter 11 Review  
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 SECTION 3 SHORT ANSWER Answer the following questions in the  
 space provided. 1. c The molar mass of a gas at STP is the  
 density of that gas (a) multiplied by the mass of 1 mol. (c)  
 multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided  
 by 22.4 L.mc06se cFMsR i-vi - Ed W. Clark High Schoolexplains  
 some of the properties of ideal gases. In this chapter, you will  
 study the predictions of kinetic-molecular theory for gases in  
 more detail. This includes the relationship among the  
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 unit area of a surface. Newton. The SI unit for force; the force  
 that will increase the speed of a 1 kg mass by 1 m/s each second  
 that the force is applied. Barometer. An instrument that  
 measures atmospheric pressure.Chemistry Chapter 11 Gases  
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 volume of 150.0 mL when its pressure is 0.947 atm. What will the  
 volume of the gas be at a pressure of 0.987 atm if theGases - Los  
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