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 SECTION 1 SHORT ANSWER 1. In order for an electron to be ejected from a metal surface, the electron must be struck by a single photon with at least the minimum energy needed to knock the electron loose.
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 This scientist proposed to explain the photoelectric effect is electromagnetic radiation in whole numbers of photons and in order for an electron to be ejected ...
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helps students learn the essential lessons associated with how electrons are arranged in atoms.
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 CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.
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 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit.
 7 Chemical Formulas and Chemical Compounds
 Modern Chemistry 28 Chapter Test Name Class Date Chapter Test A, continued ____ 8. What is the frequency of a photon whose energy is 3.4 10¹⁹ J? (h 6.626 10³⁴ J·s) ... Answers should include three of the following: All matter is composed of extremely small particles called atoms.
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questions in the space provided. 1. c In a sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit.
Stock system name such as iron(III)

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