

Acid Base Titrations Chem Worksheet 19 5 Answers

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Key Worksheet 15 Acids & Base Equilibria: Acid-Base Titrations

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Titration Acid-Base (Simple) - Practical Chemistry

An acid-base titration is a neutralization reaction that is performed in the lab in order to determine an unknown concentration (Molarity) of acid or base. As long as the concentration of one of the solutions is known, the concentration of the other reaction can be obtained through titration. In a titration reaction, the moles of acid equal the moles of base at the equivalence point.

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Acid-Base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions. The analyte (titrand) is the solution with an unknown molarity. The analyte (titrand) is the solution with an unknown molarity.

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