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# Chapter Test B

## Chemical Formulas

## And Compounds

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Chapter Test B Chemical Formulas

Chapter Test B Chemical Formulas And  
Compounds

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01 | Best way to  
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12 amu, H = 1 amu, O = 16 amu\*\* A. C 3 H 6 O 3 B. C 4 H 26 O C. C 4 H 10 O 2 D. C 5 H 14 O Question 3 A substance of phosphorus (P) and oxygen (O) is found to have a mole ratio of 0.4 moles of P for every mole of O. Chemical Formulas Practice Test Questions Chapter Test B Chemical Formulas Chapter 9 Chemical Names And Formulas Test B Answers CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO<sub>3</sub> a. copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> b. sodium sulfite (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> c. ... Chapter Test B Chemical Formulas

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#### **Step by Step**

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 rules for naming and  
 writing formulas for  
 compounds with  
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 compounds by using a  
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Chemical Formulas and

Chemical Compounds

MIXED REVIEW SHORT

ANSWER Answer the

following questions in

the space provided. 1.

Write formulas for the

following compounds:

CuCO<sub>3</sub> a. copper(II)

carbonate Na<sub>2</sub>SO<sub>3</sub> b.

sodium sulfite (NH<sub>4</sub>)

3PO<sub>4</sub> c ...

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called oxidation state,  
indicates the general  
distribution of  
electrons among the  
bonded atoms in a  
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The molecular formula of the substance is:  
 \*\*Use atomic masses of C = 12 amu, H = 1 amu, O = 16 amu\*\* A. C<sub>3</sub>H<sub>6</sub>O<sub>3</sub> B. C<sub>4</sub>H<sub>26</sub>O<sub>2</sub> C. C<sub>4</sub>H<sub>10</sub>O<sub>2</sub> D. C<sub>5</sub>H<sub>14</sub>O  
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