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Chemistry Unit 7 Rearranging Atoms

The other plan for today was to begin Unit 7 by completing the Rearranging Atoms activity. We do not have time for the nail lab, so I figured this would be a good way to introduce balancing equations. The plan was to do the Describing Chemical Reactions lab tomorrow, since today's activity seemed very straight forward.

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Chemistry Unit 7 Review

Chemistry - Unit 7 Review Chemical Reaction Model 1 Describe key characteristics of all chemical reactions, including the role of energy Explain ...

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O a. 2C 4 H 10 + 13 O 2 8 CO 2 + 10 H 2 O 2. Metal Carbonate

Metal Oxide + CO 2 a. MgCO 3 MgO + CO 2 b. Synthesis: Metal

Oxide + CO 2 Metal Carbonate 3. Metal Sulfites

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Our Current Model of Matter Conservation of Mass Chemical

Formula Subscripts in formulas Coefficient (Hint: what is the

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equation. ... A number in front of a chemical formula in an

equation that indicates how many molecules or atoms of each

reactant and product are involved in a reaction. synthesis

reaction. A reaction in which two or more substances combine to

form a new compound. A + B --> AB

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