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Given the following equation: H<sub>2</sub>(g) + F<sub>2</sub>(g) ( 2HF(g) CHAPTER 9 REVIEW Chapter menu Resources Chapter 9 Section 1 Introduction to Stoichiometry Objective • Define stoichiometry. • Describe the importance of the mole ratio in stoichiometric calculations. • Write a mole ratio relating two substances in a chemical equation. Chapter 9 Stoichiometry Table of Contents Chapter 9 - Stoichiometry. All paper copies of worksheets and notes will be provided either in class or via Google Classroom. If you lose a

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2. Given the following equation: H<sub>2</sub>(g) + F<sub>2</sub>(g) → 2HF(g)

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