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following passage. When a piece of
magnesium metal is added to dilute
hydrochloric acid, fizzing occurs and
hydrogen gas is released from the
...Date Class 6.1 Chemical Equations
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Introduction to Chemistry Exercise 2.3 -
Cations and Anions: Identify each of the
following as a cation or an anion, and
determine the charge on each. (Obj 22)
a magnesium atom with 12 protons and
10 electrons $(+12) + (-10) = +2$ This is a
 $+2$ cation. a fluorine atom with 9 protons

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Choose the correct symbol listed below
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reactions listed in questions 1-15. Also
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 $(+12) + (10) = +2$ This is a +2 cation.
a fluorine atom with 9 protons and 10
electrons

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